A New Synthesis of Pteridines

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A new synthesis of pteridines possessing a (substituted) (*Z*)-3-hydroxyprop-1-enyl group at C(6) is based on the acylation of 4-amino-5-nitrosopyrimidines with dienoic acid chlorides, followed by a highyielding intramolecular hetero-*Diels–Alder* cycloaddition and cleavage of the N–O bond leading to **4**. Thermolysis of the resulting pteridines **4** possessing a benzyloxy group at C(4) led to the products **5**, resulting from isomerisation of the 3-hydroxyprop-1-enyl to an 3-oxopropyl side chain, while the analogous pteridines **8** possessing an NH₂ group at C(4) remained unaffected.

Introduction. - Pteridines occur widely, play essential roles in growth processes and the metabolism of one-carbon units [1-4], and are in clinical use as anticancer, antiviral, antibacterial, and diuretic drugs [5]. A number of methods allow the synthesis of pteridines either from pyrimidine or from pyrazine precursors [2][3]. 4-Amino-5-nitrosopyrimidines are common, crucial intermediates in the synthesis of purines and pteridines. They were reduced to 4,5-diaminopyrimidines, as in the Traube [6] and related Pfleiderer purine syntheses [7], and in the Gabriel-Isay and Viscontini pteridine syntheses [8][9], transformed to pyrimido-oxadiazinones that react with enamines or enol ethers to form lumazines [10][11], or used directly, as in our improved modification of the Traube purine synthesis [12] and in the Timmis condensation with carbonyl compounds [13] that leads regioselectively to substituted pteridines. Recent years have seen a rapid increase in studies devoted to pteridines¹) while synthetic innovation has, with few exceptions, been directed at improving known methods [10][11][27-33]. In the context of our interest in 8-substituted guanines, we considered a new access to pteridines. The high reactivity of nitroso (NO) groups as heterodienophiles in [4+2] cycloadditions [34] should allow a facile intramolecular hetero-Diels-Alder cycloaddition of amides resulting from N-acylation of 4-amino-5-nitrosopyrimidines with penta-2,4dienoyl chlorides and lead regioselectively to C(6)- and/or C(7)-substituted pteridines. As most naturally occurring pteridines, such as folic acids, biopterin, and neopterin, as well as the anticancer drug methotrexate are substituted at C(6) [5], this would constitute a potentially useful new synthesis of pteridines.

Results and Discussion. – We examined a few conditions for the acylation with sorbyl chloride [35][36] of the poorly soluble 6-(benzyloxy)-5-nitrosopyrimidine-2,4-

¹) See, *e.g.*, [1][14–26]. A search in the *Web of Science* (January 2006) produced 352 references for the period of 2000–2006.

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a) RCH=CH-CH=CH-COCl (R=Me, H, or Ph), pyridine; 60-70% of **2a-2c** and **7**. b) Suspension in toluene, 100°; ca. 98% of **4a-4c** and **8**. c) Suspension in o-xylene, 120°; ca. 98% of **5a-5c**.

diamine (1) [37] (*Scheme*) that we already used in our synthesis of 8-substituted purines. We obtained satisfactory results, also for the acylation by penta-3,4-dienoyl [38] and 5-phenylpenta-4,5-dienoyl chloride [39], by treating a pyridine solution of 1 at -10° with a cold pyridine solution of 1.1 equiv. of the acyl chloride.

The expected amides 2a-2c were obtained as green powders. They were not stable in solution, as judged by a progressive colour change to yellow and the formation of precipitates during chromatography, or in solution, and were not characterized. The precipitates were very polar, insoluble in many organic solvents, and strongly fluorescent, suggesting a facile intramolecular [4+2] cycloaddition. The precipitates proved complex mixtures. They were not separated. Fortunately, heating the crude acylation products 2a-2c in toluene under reflux led almost quantitatively to the pure pteridines **4a**–**4c**. The products **3a**–**3c** of the expected [4+2] cycloaddition appear to undergo a rapid eliminative cleavage of the weak N–O bond²), followed by tautomerisation.

The 5-nitrosopyrimidine-2,4,6-triamine (6) [41] was similarly acylated with sorbyl chloride. Heating the resulting amide 7 in toluene induced a similar, more sluggish transformation, leading to the diaminopteridine 8, again in high yields, as precipitates that were purified by washing with H_2O , AcOEt, and Et₂O.

Remarkably, continued heating of $4a-4c^3$), best performed in boiling *o*-xylene, led in high yields to the ketones $5a-5c^3$). The transformation of 4 to 5 is rationalised by a [1,5]-sigmatropic H shift, followed by tautomerisation of the intermediate enol enamine to the imino ketones 5. A precedent for such a transformation is found in the tautomerisation reported by *Coppola* and *Damon* [42]. The diaminopteridine 8 did not undergo such a transformation. Longer heating or higher temperatures led to its progressive decomposition.

The structure of the pteridine derivatives 4 and 8 follows unambiguously from their spectroscopic data. Thus, the constitution of the secondary alcohol 4a is evidenced by high-resolution MALDI-MS, elemental analysis, IR bands at 3420 cm⁻¹ (OH; NH bands at 3323 and 3208 cm⁻¹), a comparison of their UV spectra with those of related pteridines [27] [43], and the ¹H- and ¹³C-NMR data (*Tables 1 and 2* in the *Exper. Part*). No 13 C-NMR spectra of pteridin-7(8H)-ones were found by a Beilstein X-fire search. The assignment of the ¹³C ss of **4** and **5** to the six quaternary C-atoms was first based on the predictions of ChemDraw Ultra V. 9 and on a comparison, for C(2) and C(4), with the C(2) and C(6) signals of O^6 -methylguanosines [44] and then revised – for the assignment of C(2) and C(7) - on the basis of an HMBC spectrum of 4a. It shows cross-peaks between the two ds of PhCH₂ and the s at 164.51 ppm (C(4)), between the s of NH₂ and the ss at 164.51, 161.39 (C(8a)), and 150.62 ppm (C(2)), and between the dd of H-C(1') and the ss at 157.13 ppm (C(7)) and 146.08 (C(6)). J(1',2') of the olefinic H-atom of 12.3 Hz is in keeping with the (Z)-configuration as evidenced by a comparison with the data of related 2-(3-hydroxyprop-1-envl)pyridines [45]. The structure of $\mathbf{8}$ is evidenced by the high-resolution MALDI mass spectrum, an ATR-IR OH band at 3460 cm⁻¹, and broader NH bands at 3327 and 3175 cm⁻¹, and the NMR spectra ((D_6)DMSO) where the Me group resonates at 1.23 ppm as a d (J=6.3 Hz), the olefinic H-atom at 6.56 (dd, J=12.0 and 1.2 Hz) and 5.87 ppm (dd, J=12.0 and 7.8 Hz), and the allylic H-atom at 5.15 ppm (m), in keeping with a ¹³C q at 22.43 ppm, and ds at 118.76, 144.01, and 63.07 ppm.

Also the structure of the ketones 5 follows unambiguously from their analytical data. Thus, the 3-oxobutyl substituent of 5a is evidenced by the disappearance of the

³) In contact with air, the aldehyde **5b** is easily oxidised to the corresponding (*E*)-configured α,β -unsaturated aldehyde **5d** that was also formed in small amounts from **4b**.



²) See [40] for a base-promoted eliminative cleavage of a N-O bond.

signals corresponding to the olefinic C=C bond of **4a**, a Me *s* at 2.07 ppm, and two *triplet*-like signals of an isolated ethylene group at 2.81 and 2.71 ppm. The PhC H_2 of **5a** resonate as a *s* at 5.43 ppm (¹³C-NMR: *t* at 67.42 ppm), while PhC H_2 of **4a** resonates as two *ds* at 5.49 and 5.44 ppm. The C=O group of **5a** is evidenced by a ¹³C *s* at 207.77 ppm and an IR band at 1708 cm⁻¹. The ATR-IR OH/NH band of **4a** at 3420 cm⁻¹ is replaced by a much weaker NH band at 3427 cm⁻¹ for **5a**.

This synthesis of pteridines possessing a configurationally defined 3-hydroxyprop-1enyl substituent at C(6) should allow a convenient access to naturally occurring pteridines and their analogues; work towards this goal is in progress.

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Experimental Part

General. See [46]. UV Spectra: MeOH, λ_{max} (log ε).

General Procedure for the Preparation of the Pteridin-7-ones 4a-4c and 8. A soln. of 1 [37] (245 mg, 1.0 mmol) or 6 [41] (154 mg, 1.0 mmol) in dry pyridine (10 ml) was cooled to -10° , treated with the 2,4-dienoyl chloride (1.1 mmol), and stirred for 12 h. The soln. was diluted with CH₂Cl₂ (50 ml), washed with brine (3×20 ml), dried (MgSO₄), and evaporated. FC (CH₂Cl₂/MeOH 100:1) of the blue residue gave green powders of 2a-2c and 7 (60–70%), resp. A suspension of 2a-2c and 7 (1 mmol) in toluene was heated for 3–5 h to 100°, when a yellow suspension was formed. After evaporation, the solid yellow residue was washed with H₂O, AcOEt, and Et₂O. Drying of the residue *in vacuo* (*i.v.*) gave 4a-4c and 8 (*ca.* 98%), resp. Yellow powders.

General Procedure for the Transformation of 4a-4c to the Carbonyl Compounds 5a-5c. A suspension of 4a-4c (1 mmol) in *o*-xylene was degassed, heated for 12 h to 120°, and evaporated. The solid yellow residue was washed with H₂O and Et₂O. Drying of the residue *i.v.* gave 5a-5c (*ca.* 98%). Yellow powders.

2-Amino-6-[(Z)-3-hydroxybut-1-enyl]-4-(phenylmethoxy)pteridin-7(8H)-one (4a). M.p. 242° (dec.). UV: 376 (4.18), 285 (3.84), 208 (4.46). IR (ATR): 3420m, 3323w, 3208m, 2834w, 2737w (br.), 1802w, 1670w, 1614s, 1560s, 1538m, 1496m, 1490m, 1464m, 1428s, 1387m, 1356s, 1327m, 1307m, 1182s, 1052s, 975m, 927m, 905m. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 7.54–7.31 (*m*, 5 arom. H); 5.49, 5.44 (2*d*, J=12.6, PhCH₂). ¹³C-NMR ((D₆)DMSO, 100 MHz; assignments based on a HSQC and a HMBC spectrum): see *Table 2*; additionally, 136.31 (*s*); 128.23 (2*d*); 127.77 (*d*); 127.36 (2*d*); 67.32 (*t*, PhCH₂). HR-MALDI-MS: 362.1223 (82, [M+Na]⁺, C₁₇H₁₇N₅NaO₃⁺; calc. 362.1229), 340.1402 (35, [M+H]⁺, C₁₇H₁₈N₅O₃⁺; calc. 340.1410), 322.1299 (100, [M-OH]⁺, C₁₇H₁₆N₅O₂⁺; calc. 322.1304). Anal. calc. for C₁₇H₁₇N₅O₃ (339.35): C 60.17, H 5.05, N 20.64; found: C 60.38, H 5.15, N 20.50.

2-*Amino-6-[(Z)-3-hydroxyprop-1-enyl]-4-(phenylmethoxy)pteridin-7(8*H)-*one* (4b). M.p. 220° (dec.). UV: 360 (3.91), 287 (3.70), 216 (4.43). IR (ATR): 3367*m*, 3330*m*, 3193*m*, 2888*w*, 2832*w*, 2766*w* (br.), 1673*m*, 1650*s*, 1610*s*, 1561*s*, 1534*s*, 1489*s*, 1440*s*, 1388*m*, 1343*s*, 1325*m*, 1265*m*, 1189*s*, 1086*m*, 1064*m*, 1037*m*, 1016*s*, 950*m*, 913*s*. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 7.50–7.31 (*m*, 5 arom. H); 5.49 (*s*, PhCH₂); weak signals at 9.69 (*d*, J=7.8), 7.68 (*d*, J=16.2), and 7.12 (*dd*, J=16.2, 7.8) denote a contamination by 5 to 10% of 5d. ¹³C-NMR ((D₆)DMSO, 100 MHz): see *Table 2*; additionally, 136.58 (*s*); 128.42 (2*d*); 127.78 (*d*); 127.30 (2*d*); 67.10 (*t*, PhCH₂). HR-MALDI-MS: 348.1066 (43, [M+Na]⁺, C₁₆H₁₅N₅NaO⁺₃; calc. 348.1073), 326.1247 (56, [M+H]⁺, C₁₆H₁₆N₅O⁺₃; calc. 326.1253), 308.1140 (100, [M-OH]⁺, C₁₆H₁₄N₅O⁺₂; calc. 308.1147).

2-Amino-6-[(Z)-3-hydroxy-3-phenylprop-1-enyl]-4-(phenylmethoxy)pteridin-7(8H)-one (4c). M.p. 205° (dec.). UV: 381 (4.16), 280 (3.91), 233 (4.22), 211 (4.55). IR (ATR): 3422w, 3323w, 3206m, 2734w (br.), 1662m, 1613s, 1561s, 1487s, 1432s, 1394m, 1354s, 1326m, 1264m, 1178m, 1090m, 1046m, 1002m, 938m, 908m. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 12.46 (s, HN(8)); 7.58–7.06

Table 1. Selected ¹ H-NMR Chemical Shifts [ppm] and Coupling Constants [Hz] of the Pteridin-7(8H)-ones
$4a-4c$, $5a-5c$, and 8 in $(D_6)DMSO$

		-~	40	58	50	5C	8
H-N(8)	12.40	12.38	12.46	12.34	12.37	12.35	12.12
$H_2N-C(2)$	7.24	7.20	a)	7.08	7.09	7.06	7.29 ^b)
H–C(1')	6.61	6.68	6.68	2.81	2.90	2.99	6.56
H–C(2')	5.95	6.09	6.02	2.71	2.76	3.40	5.87
H–C(3')	5.15	4.53	6.36	_	9.73	_	5.15
HO–C(3')	4.89	4.92	5.48	_	_	_	4.81
Me(4')	1.20	_	_	2.07	_	_	1.23
J(1',2')	12.3	12.0	12.0	6.3	6.6	6.6	12.0
J(1',3')	1.2	2.4	0	-	-	-	1.2
J(2',3')	7.5	5.4	9.0	-	1.2	-	7.8
J(3',OH)	4.2	c)	4.2	-	-	-	4.5
J(3',4')	6.3	-	-	-	-	-	6.3

Table 2. Selected ¹³C-NMR Chemical Shifts [ppm] of the Pteridin-7(8H)-ones 4a-4c, 5a-5c, and 8 in $(D_6)DMSO$

	4a ^a)	4b	4c	5a	5b	5c ^b)	8 ^b)
C(2)	150.62	150.65	151.01	151.92	151.72	152.03	148.88
C(4)	164.51	164.53	164.61	164.25	164.27	164.14	160.63
C(4a)	107.54	107.71	107.56	106.70	106.82	106.71	106.49
C(6)	146.08	146.52	145.83	151.17	151.27	151.13	143.68
C(7)	157.13	157.17	157.26	157.02	157.03	157.01	157.79
C(8a)	161.39	161.37	161.41	161.17	161.18	161.06	162.20
C(1′)	118.56	119.18	118.50	26.64	25.17	26.44	118.76
C(2')	145.75	142.63	142.50	38.25	39.18	34.17	144.01
C(3')	63.45	60.28	67.45	207.77	202.53	199.01	63.07
C(4')	22.26	_	_	30.16	_	_	22.43

(*m*, 10 arom. H, NH₂); 5.64, 5.43 (2*d*, J = 12.6, PhCH₂). ¹³C-NMR ((D₆)DMSO, 100 MHz): see *Table* 2; additionally, 144.33, 136.22 (2*s*); 128.41 (2*d*); 128.16 (2*d*); 127.77 (*d*); 127.70 (2*d*); 126.50 (*d*); 126.44 (2*d*); 67.52 (*t*, PhCH₂). HR-MALDI-MS: 424.1376 (40, [M+Na]⁺, C₂₂H₁₉N₅NaO₃⁺; calc. 424.1386), 402.1555 (45, [M+H]⁺, C₂₂H₂₀N₅O₃⁺; calc. 402.1566), 384.1450 (100, [M-OH]⁺, C₂₂H₁₈N₅O₂⁺; calc. 384.1460).

2-Amino-6-(3-oxobutyl)-4-(phenylmethoxy)pteridin-7(8H)-one (**5a**). M.p. 245° (dec.). UV: 343 (4.08), 284 (3.78), 212 (4.40). IR (ATR): 3427w, 3322w, 3202w, 2900w, 2835w, 2760w (br.), 1708w, 1666m, 1615s, 1557s, 1498m, 1431s, 1391m, 1354s, 1307m, 1253m, 1178m, 1087m, 1058m, 978m, 930m, 911m. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 7.51–7.32 (m, 5 arom. H); 5.44 (s, PhCH₂). ¹³C-NMR ((D₆)DMSO, 100 MHz): see *Table 2*; additionally, 136.34 (s); 128.42 (2d); 128.38 (2d); 128.10 (d); 67.42 (t, PhCH₂). HR-MALDI-MS: 362.1222 (49, $[M+Na]^+$, $C_{17}H_{17}N_5NaO_3^+$; calc. 362.1229), 340.1398 (100, $[M+H]^+$, $C_{17}H_{18}N_5O_3^+$; calc. 340.1410).

2-Amino-6-(3-oxopropyl)-4-(phenylmethoxy)pteridin-7(8H)-one (**5b**). M.p. 210° (dec.). UV: 345 (3.89), 282 (3.74), 209 (4.37). ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 7.49–7.31

1144

(m, 5 arom. H); 5.43 $(s, \text{PhC}H_2)$. ¹³C-NMR $((D_6)$ DMSO, 100 MHz): see *Table 2*; additionally, 136.46 (s); 128.41 (2*d*); 128.12 (2*d*); 127.98 (*d*); 67.20 (*t*, PhCH₂). HR-MALDI-MS: 348.1067 (31, $[M+\text{Na}]^+$, $C_{16}H_{15}N_5\text{NaO}_3^+$; calc. 348.1073), 326.1249 (79, $[M+\text{H}]^+$, $C_{16}H_{16}N_5\text{O}_3^+$; calc. 326.1253).

2-Amino-6-(3-oxo-3-phenylpropyl)-4-(phenylmethoxy)pteridin-7(8H)-one (**5c**). M.p. 255° (dec.). UV: 343 (3.57), 279 (3.35), 235 (3.75), 204 (4.19). IR (ATR): 3414w, 3322w, 3204w, 2916w, 2831w, 2736w (br.), 1684m, 1662m, 1644s, 1619s, 1559s, 1511m, 1497m, 1433s, 1392m, 1355s, 1298s, 1205m, 1183s, 1088m, 1055s, 987m, 920m. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*; additionally, 7.98–7.29 (m, 10 arom. H); 5.38 (s, PhCH₂). ¹³C-NMR ((D₆)DMSO, 125 MHz; assignments based on HSQC spectrum): see *Table 2*; additionally, 136.81, 136.33 (2s); 132.83 (d); 128.51 (2d); 128.21 (2d); 127.78 (3d); 127.72 (2d); 66.96 (t, PhCH₂). HR-MALDI-MS: 424.1380 (59, $[M+Na]^+$, $C_{22}H_{19}N_5NaO_3^+$; calc. 424.1386), 402.1563 (100, $[M+H]^+$, $C_{22}H_{29}N_5O_3^+$; calc. 402.1566).

2,4-Diamino-6-[(Z)-3-hydroxybut-1-enyl)pteridin-7(8H)-one (8). M.p. 210° (dec.). IR (ATR): 3460w, 3327m, 3175m, 2969w, 2924w, 2847w, 2721w (br.), 1826w, 1615s, 1556s, 1530s, 1480s, 1454s, 1401s, 1323m, 1257m, 1153m, 1108m, 1048s, 930m, 903m. ¹H-NMR ((D₆)DMSO, 300 MHz): see *Table 1*. ¹³C-NMR ((D₆)DMSO, 100 MHz; assignments based on HSQC spectrum): see *Table 2*. HR-MALDI-MS: 271.0914 (26, $[M+Na]^+$, $C_{10}H_{12}N_6NaO_2^+$; calc. 271.0919), 249.1093 (37, $[M+H]^+$, $C_{10}H_{13}N_6O_2^+$; calc. 249.1100), 231.0991 (100, $[M-OH]^+$, $C_{10}H_{11}N_6O^+$; calc. 231.0994).

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